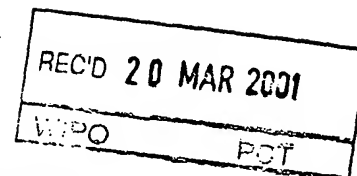


PATENT COOPERATION TREATY

PCT

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

(PCT Article 36 and Rule 70)



14

Applicant's or agent's file reference 91830/667	FOR FURTHER ACTION See Notification of Transmittal of International Preliminary Examination Report (Form PCT/IPEA/416)	
International application No. PCT/US00/16712	International filing date (day/month/year) 16 JUNE 2000	Priority date (day/month/year) 16 JUNE 1999
International Patent Classification (IPC) or national classification and IPC IPC(7): C12P 19/34; C12N 9/00, 13/00 and US Cl.: 435/91.1, 173.7, 183		
Applicant UNIVERSITY OF CINCINNATI		

<p>1. This international preliminary examination report has been prepared by this International Preliminary Examining Authority and is transmitted to the applicant according to Article 36.</p> <p>2. This REPORT consists of a total of <u>4</u> sheets.</p> <p><input type="checkbox"/> This report is also accompanied by ANNEXES, i.e., sheets of the description, claims and/or drawings which have been amended and are the basis for this report and/or sheets containing rectifications made before this Authority. (see Rule 70.16 and Section 607 of the Administrative Instructions under the PCT).</p> <p>These annexes consist of a total of <u>0</u> sheets.</p>
<p>3. This report contains indications relating to the following items:</p> <p>I <input checked="" type="checkbox"/> Basis of the report</p> <p>II <input type="checkbox"/> Priority</p> <p>III <input type="checkbox"/> Non-establishment of report with regard to novelty, inventive step or industrial applicability</p> <p>IV <input type="checkbox"/> Lack of unity of invention</p> <p>V <input checked="" type="checkbox"/> Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement</p> <p>VI <input type="checkbox"/> Certain documents cited</p> <p>VII <input type="checkbox"/> Certain defects in the international application</p> <p>VIII <input type="checkbox"/> Certain observations on the international application</p>

Date of submission of the demand 06 DECEMBER 2000	Date of completion of this report 19 FEBRUARY 2001
Name and mailing address of the IPEA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231 Facsimile No. (703) 305-3230	Authorized officer TERRY J. DEY GERALD G. LEFFERS PARALEGAL SPECIALIST TECHNOLOGY CENTER 1600 Telephone No. (703) 308-0196

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/US00/16712

I. Basis of the report

1. With regard to the elements of the international application:*

☒ the international application as originally filed☒ the description:

pages 1-32 , as originally filed
pages NONE , filed with the demand
pages NONE , filed with the letter of _____

☒ the claims:

pages 33-41 , as originally filed
pages NONE , as amended (together with any statement) under Article 19
pages NONE , filed with the demand
pages NONE , filed with the letter of _____

☒ the drawings:

pages NONE , as originally filed
pages NONE , filed with the demand
pages NONE , filed with the letter of _____

☒ the sequence listing part of the description:

pages NONE , as originally filed
pages NONE , filed with the demand
pages NONE , filed with the letter of _____

2. With regard to the **language**, all the elements marked above were available or furnished to this Authority in the language in which the international application was filed, unless otherwise indicated under this item.

These elements were available or furnished to this Authority in the following language _____ which is:

- ☐ the language of a translation furnished for the purposes of international search (under Rule 23.1(b)).
☐ the language of publication of the international application (under Rule 48.3(b)).
☐ the language of the translation furnished for the purposes of international preliminary examination (under Rules 55.2 and/or 55.3).

3. With regard to any **nucleotide and/or amino acid sequence** disclosed in the international application, the international preliminary examination was carried out on the basis of the sequence listing:

- ☐ contained in the international application in printed form.
☐ filed together with the international application in computer readable form.
☐ furnished subsequently to this Authority in written form.
☐ furnished subsequently to this Authority in computer readable form.
☐ The statement that the subsequently furnished written sequence listing does not go beyond the disclosure in the international application as filed has been furnished.
☐ The statement that the information recorded in computer readable form is identical to the written sequence listing has been furnished.

4. ☒ The amendments have resulted in the cancellation of:

☒ the description, pages NONE
☒ the claims, Nos. NONE
☒ the drawings, sheets/fig NONE

5. ☐ This report has been drawn as if (some of) the amendments had not been made, since they have been considered to go beyond the disclosure as filed, as indicated in the Supplemental Box (Rule 70.2(c)).**

* Replacement sheets which have been furnished to the receiving Office in response to an invitation under Article 14 are referred to in this report as "originally filed" and are not annexed to this report since they do not contain amendments (Rules 70.16 and 70.17).

**Any replacement sheet containing such amendments must be referred to under item 1 and annexed to this report.

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/US00/16712

V. Reasoned statement under Article 35(2) with regard to novelty, inventive step or industrial applicability; citations and explanations supporting such statement

1. statement

Novelty (N)

Claims 1-51 YESClaims NONE NO

Inventive Step (IS)

Claims 1-51 YESClaims NONE NO

Industrial Applicability (IA)

Claims 1-51 YESClaims NONE NO

2. citations and explanations (Rule 70.7)

ClaimS 1-51 the criteria set out in PCT Article 33(2)-(4), because the prior art does not teach or fairly suggest the claimed methods of isolating extrachromosomal DNA featuring the use of chaotropic agents to precipitate the extrachromal DNAs.

----- NEW CITATIONS -----

NONE

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/US00/16712

Supplemental Box

(To be used when the space in any of the preceding boxes is not sufficient)

Continuation of: Boxes I - VIII

Sheet 10

AGENT AND PROCESS FOR ISOLATION OF EXTRA-CHROMOSOMAL NUCLEIC ACIDS

5

Field of the Invention

The invention relates to compositions and methods for isolating extra-chromosomal
10 nucleic acids from biological materials. More particularly, the invention relates to
plasmid DNA isolation methods employing non-toxic chaotropic agents.

Background of the Invention

Propagation and isolation of extra-chromosomal DNA is a crucial process in
15 genetic cloning and engineering. Extra-chromosomal DNA is distinct from
chromosomal or genomic DNA that forms the chromosomes. In eukaryotic and
prokaryotic cells, genomic DNA is attached or bound to a membrane. In eukaryotes,
such as mammalian or yeast cells, the genomic DNA is contained within and bound to
the nuclear membrane. In prokaryotes, such as bacteria, genomic DNA is bound to
20 the plasma membrane surrounding the cytoplasm. While all cells contain genomic
DNA, only some have extra-chromosomal DNA. As the name implies, extra-
chromosomal DNA is not attached to the chromosomal (genomic) DNA, and as such,
is not attached to cell membranes. Extra-chromosomal DNA may be naturally
occurring or artificially introduced into eukaryotic or prokaryotic cells. Extra-

chromosomal DNA forms known and used in genetic engineering or cloning include, but are not limited to, plasmid, cosmid, bacterial artificial chromosome (BAC), and yeast artificial chromosome (YAC) DNA.

In order to distinguish the methods of the present invention from the prior art, it is helpful to understand the general process involved in isolation of DNA. Of particular importance is an appreciation for the differences between two uses of the term "lysis": one for the isolation of genomic DNA, the other for extra-chromosomal DNA. In the first case, cells or bacteria are lysed to harvest genomic DNA by completely disrupting all cell membranes. These membranes include the plasma membranes, which form the outer wall of the cell, and the intracellular membranes, including the nuclear membrane. In this manner, genomic DNA is released from its membrane-bound state into the lysate. In the second case, cells or bacteria are treated with an alkaline lysis, which is a more gentle and incomplete disruption of the cell membranes. With an alkaline lysis, genomic DNA remains tethered to the membranes, while extra-chromosomal DNA is released into the lysate. Sedimentation of the cellular debris into a pellet effectively removes genomic DNA from the lysate, since it remains bound to the cellular debris. In each case, RNA and a multitude of protein contaminants are also released into the lysates and must be removed. A variety of purification steps must be performed on the crude lysates to obtain isolates of the DNA of interest.

There are many well-known and widely-used methods for isolation of DNA from biological material. Biological materials can include homogenized tissue samples, yeast or bacterial cultures, tissue culture cells originating from complex

organisms such as mammals or insects but subsequently grown *in vitro*, and agarose gel containing semi-pure DNA preparations.

Methods in the prior art of extra-chromosomal DNA isolation rely upon the use of toxic or caustic agents for organic extraction of impurities, or employ high affinity columns, resins, or beads to bind extra-chromosomal DNA contained within the partially processed crude biological material prior to the washing and elution or recovery steps. Examples of conventional methods can be found in *Molecular Cloning, A Laboratory Manual* Second Edition. Sambrook, J.; Fritsch, E.F.; Maniatis, T. Cold Spring Harbor Laboratory Press, 1989, or *Current Protocols in Molecular Biology, Volume 1*, John Wiley & Sons, Inc., 1995. Conventional methods require the use of toxic chemicals, such as phenol:chloroform mixtures. Aside from safety issues, the toxic chemicals can affect quality of the DNA isolated, i.e. the DNA macromolecules may be damaged or degraded. Another disadvantage with conventional methods is that they are generally more time-consuming than those using a DNA-binding matrix. Conventional methods can take up to several days and typically require the use of expensive ultracentrifugation equipment. Commercially available kits using a binding matrix take only a few hours, but are more expensive and typically result in lower yields of DNA than with conventional methods. Examples of methods using DNA-binding materials include the Qiagen MAXI-PREP and Promega WIZARD PUREFECTION PREP kits. With either conventional methods or use of DNA-binding matrices, purity of the isolate is frequently less than optimal. Typically, the cleanest (and therefore, most useful) DNA is isolated at the expense of the quantity of DNA that can be recovered.

The present invention discloses materials and methods used in an unexpected sequence to produce high yields of optimally pure extra-chromosomal DNA. The need for toxic chemicals or DNA-binding materials is eliminated. The process can be completed in about 2-3 hours using equipment commonly available in any molecular biology laboratory, using a chaotropic agent specifically formulated for the purification of extra-chromosomal DNA.

Chaotropic agents similar to the ones used in the present invention have been previously used only in the isolation of genomic DNA. Homogenized tissue samples or cell suspensions are treated with chaotropic agents, such as those disclosed in the published application W097/05248, to lyse the cell membranes and release genomic DNA into the lysate. If such materials are used to harvest extra-chromosomal DNA, the resultant release of genomic DNA contaminates the extra-chromosomal DNA. In fact, others have attempted to use such materials for isolating plasmid DNA, but have failed to devise the needed modifications to the materials and methods. Therefore, there are no prior art references teaching the use of chaotropic agents for plasmid or other extra-chromosomal DNA preparations.

In the present invention, a chaotropic solution has been modified specifically for the isolation of extra-chromosomal DNA. However, the process of the invention does not include lysis of cell homogenates or bacterial suspensions with chaotropic agents. The cells or bacteria are lysed using a standard alkaline lysis method. The lysate is separated from the cellular debris and genomic DNA, and then treated with the chaotropic agents as a purification step. The yield of extra-chromosomal DNA isolated using the present invention materials and method is significantly greater and purer than those using conventional methods. The time of isolation using the present

invention method is comparable to those dependent upon the use of DNA binding matrices. The yield is also enhanced because the method of the present invention is not influenced by the electrostatic, temperature and topological variables inherent in DNA-binding resin isolation procedures. Since the present invention requires only aqueous solutions of relatively inexpensive chemicals, the cost is significantly less than other methods. Perhaps the greatest benefit of the present invention is that the extra-chromosomal DNA is isolated in an undegraded and intact condition, providing a more useful final product.

10 **Brief Description of the Invention**

The present invention provides methods and materials for isolating substantially pure and undegraded extra-chromosomal nucleic acids from biological materials. The solution contains effective amounts of a chaotropic agent, a buffer present in an amount sufficient to maintain an alkaline pH equal to or greater than about 7.5, salt, detergent, and alcohol. The invention is also a method for isolating substantially pure and undegraded extra-chromosomal DNA from biological materials. The steps in the method include precipitating extra-chromosomal DNA and contaminating RNA from a cell lysate, treating the extra-chromosomal DNA with RNase to digest the contaminating RNA, removing residual contaminants from extra-chromosomal DNA by adding the chaotropic solution, precipitating the DNA with an alcohol, pelleting the DNA by sedimentation, and washing the sedimented DNA with alcohol. An alternative use of the invention is for recovery of any type of DNA from agarose gel preparations.

Detailed Description of the Invention

The following terms have the following meanings for the purposes of this application, unless expressly stated to the contrary herein.

By "lysis" is meant cellular dissociation involving physical disruption and
5 breakage of the cell wall and/or membrane, causing intracellular components including nucleic acids to be released into the surrounding medium.

By "incomplete lysis" is meant the lysis of a cell or a group of cells which renders the cell chromosomes, cell membranes and other components capable of forming a precipitate while leaving the extracellular DNA in solution.

10 By "alkaline lysis" is meant the breaking open of a cell or a group of cells, or the liberation of some or all of the intracellular components following treatment of the cell or cells with an alkali or base which partially digests the organism's cell wall or membrane. This type of lysis breaks open a cell or a group of cells, and renders the cell chromosomes, cell membranes and other components capable of forming a
15 precipitate while leaving the extracellular DNA in solution.

By "enzymatic lysis" is meant the breaking open of a cell or a group of cells, or the liberation of some or all of the intracellular components following treatment of the cell or cells with an enzyme which partially digests the organism's cell wall. This type of lysis breaks open a cell or a group of cells, and renders the cell chromosomes,
20 cell membranes and other components capable of forming a precipitate while leaving the extracellular DNA in solution.

By "detergent" or "surfactant" is meant a molecule or class of molecules which have a hydrophobic region or moiety capable of interacting with hydrophobic solvents and the hydrophobic portions of cellular membranes, and a hydrophilic

region or moiety which may have a positive or a negative charge in solution, or alternately may have a polar region with no charge at all.

Reference to the "release of nucleic acids" is intended to mean the liberation of nucleic acids in sufficient quantities such that the method of release is useful for
5 further purification and subsequent use in molecular biology.

By "nucleic acid" or "nucleic acids" is meant polydeoxyribonucleotides or polyribonucleotides of at least two, and preferably 10 or more nucleotides in length. The term "nucleic acid" includes polynucleotides, oligonucleotides, and DNA or RNA molecules. The term "nucleic acid" can refer to either single-stranded or double-
10 stranded polynucleotides, or both.

By "biological sample" is meant any specimen or sample containing substances of biological or biochemical origin. "Biological materials" include, but are not limited to, microorganisms such as bacteria transformed with recombinant bacteriophage, plasmid or cosmid DNA or BACs; yeast transformed with yeast
15 expression vectors or YACs; insect cell systems infected with viral expression vectors (*e.g.* baculovirus); plant cell systems transformed with viral expression vectors (*e.g.* cauliflower mosaic virus, CaMV; tobacco mosaic virus, TMV), or cells derived from higher organisms such as mammalian tissue culture cells transfected with recombinant plasmid DNA or infected with viral expression vectors. "Biological material" may
20 also refer to agarose or polyacrilamide gel preparations used to separate any nucleic acids by size, charge or other characteristic.

As used herein, "substantially purified" refers to recovery of nucleic acid which is at least 80% and preferably 90-95% purified with respect to removal of a contaminant, *e.g.*, cellular components such as protein, lipid or salt; thus, the term

“substantially purified” generally refers to separation of a majority of cellular proteins or reaction contaminants from the sample, so that compounds capable of interfering with the subsequent use of the isolated nucleic acid are removed.

As used herein, “buffer” refers to a buffered solution that resists changes in pH
5 by the action of its acid-base conjugate components.

As used herein, “chaotropic environment” refers to an environment which contains appropriate chaotropic agents, such as urea in sufficient concentration to disrupt the tertiary structure of proteins, or which is maintained at a temperature or other condition which causes such disruption.

10 Chaotropic agents or conditions such as temperature and pH may disrupt structure in a variety of ways, including the disruption of hydrogen bonds. Suitable chaotropic environments include 2-8M urea, 4-7M guanidinium, detergents such as SDS at concentrations around 0.1% by weight, and acids such as acetic acid at concentrations of about 1M, basic conditions of, *e.g.*, pH 11 and above, and elevated
15 temperatures. When placed in a chaotropic environment, the normal physiological conformation of proteins may be reversibly as well as irreversibly altered, and the primary structure may be “unfolded” to varying degrees, depending on the concentration of the chaotropic agent and the degree of severity of other chaotropic conditions. It should be understood that agents and/or conditions that create
20 chaotropic environments can be used in combination or in sequence. For example, mixtures of chaotropic agents can be used, or the biological sample may first be placed in a chaotropic environment created by one chaotropic agent, and then subjected to a second chaotropic environment created by another agent or by temperature.

The products and methods of the present invention provide a highly efficient, simple means of isolating DNA from a single biological sample, such as bacteria. Advantageously, these results can be achieved without the use of toxic or caustic reagents, DNA-binding matrices, or the use of expensive ultracentrifugation
5 equipment.

In its broadest aspects, the invention encompasses solution and methods for isolating substantially pure and undegraded extra-chromosomal nucleic acids from biological materials.

The present invention is an entirely liquid based system that will not require
10 the production or use of columns or beads employing DNA-binding materials. No toxic chemicals such as phenol or chloroform will be needed. The use of the present invention to isolate plasmid or other extra-chromosomal DNA is significantly less expensive and as fast or faster than methods currently known and available. The methods can be adapted to differing amounts of biological materials by changing
15 volumes of the reagents used. For example, plasmid DNA can be harvested from bacterial cultures in what are conventionally known as "mini-", "midi-", or "maxi-prep" amounts. The extra-chromosomal DNA isolated using the materials and methods of this invention is of particularly high purity, quality, and thus, usefulness.

The present invention is distinct from the prior art disclosures of chaotropic
20 agents used to purify DNA because those materials and methods were designed for the isolation of genomic DNA. As such they use a formulation and method that will disrupt cell membranes and release both genomic and any extra-chromosomal DNA into a heterogeneous mixture. Isolation of extra-chromosomal DNA from such a mixture is, at best, an impractical and highly inefficient process.

The present invention is also distinct from those in the prior art that are directed to the isolation of extrachromosomal DNA. In DNA-binding matrix methods and conventional methods, cells or bacteria are first lysed using a standard alkaline lysis method. Alkaline lysis releases extrachromosomal-DNA and RNA from the cells or bacteria into the crude lysate supernatant, but genomic DNA remains
5 membrane-bound and is pelleted in the bacterial debris along with the membranes with centrifugation. RNA is digested by adding RNase to the lysate. In methods using DNA-binding matrices, the RNase-treated lysate is applied to column or beads containing the DNA-binding materials, washed to remove protein contaminants, and
10 eluted. In conventional methods, the DNA in a lysate is separated from contaminants first by ultracentrifugation on a gradient, such as cesium salts, and then dialyzed and/or purified with a series of phenol:chloroform extractions.

In the present invention, a lysate is first prepared using a standard alkaline lysis method. The neutralized solution is treated with an alcohol precipitation
15 followed by RNase to degrade contaminating RNA in the lysate. Next, a chaotropic solution is added to the lysate to remove residual protein and degraded RNA contaminants from the extra-chromosomal DNA.

The steps of the present methods are described more fully as follows:

20 **1. Cell lysis**

As already mentioned a typical method of the present invention involves the recovery of purified plasmid from a bacterial lysate. Accordingly, typical purification procedures first require the preparation of the bacterial lysate from a bacterial culture. Those skilled in the art are familiar with methods for lysing bacteria. The present

invention is described in terms of the known alkaline lysis procedure, however, any procedure which effectively lysis the bacterial cells and releases plasmid DNA is applicable, such as enzymatic lysis. Typically these procedures involve adding a bacterial pellet to solutions which break open the bacterial cells, and cause the
5 bacterial chromosomes, cell membranes and other components to form a precipitate while leaving the plasmid DNA in solution.

The alkaline lysis is generally prepared by pelleting a suspension of a biological material, such as a liquid culture of bacteria, through centrifugation. The supernatant is poured off and the pellet resuspended in buffer solution. The biological
10 material is then lysed by addition of a strong base, *e.g.*, sodium hydroxide, in the presence of a detergent such as sodium dodecyl sulfate (SDS). The alkaline lysis is terminated by the addition of a neutralization solution such as sodium or potassium acetate. The neutralized lysis mixture is centrifuged to pellet all cellular debris, including membranes and membrane-bound chromosomal DNA. The lysate can
15 alternatively be separated from the cellular debris by various filtration methods well known in the art, *e.g.*, by gravity or by vacuum. The supernatant ("lysate") is poured into a clean tube for the precipitation step.

2. Precipitation and RNase treatment

Alcohol, such as ethanol, is added to the lysate for a final concentration of
20 about 70%. After mixing thoroughly, the solution is centrifuged to pellet the nucleic acids, which will include RNA and extra-chromosomal DNA. The supernatant is discarded. The pellet is resuspended in a buffer solution containing RNase to degrade all RNA contaminants. The concentration of the buffer solution is generally an amount such that the pH is from about 7 to about 12, preferably where the pH is from

about 7 to about 10, and most preferably where the pH is from about 7 to about 8. The RNase incubation is generally performed at 37° (but can be performed at room temperature) for about 10 minutes.

3. Chaotropic Solution Treatment

5 The RNase-treated solution is then mixed with a chaotropic solution and stored at about room temperature for about 10 minutes. The mixture of the RNase-treated solution and the chaotropic solution is then treated with a solvent (0.3 to 1.0 of solution volume) in order to precipitate the extra-chromosomal DNA, and subsequently centrifuged to pellet the DNA away from contaminating proteins and
10 degraded RNA. The supernatant is removed and discarded. The DNA pellet is washed with a solvent (about 60% to about 80%) to remove any residual traces of the chaotropic solution. Following centrifugation, the supernatant is removed and the DNA pellet is allowed to dry. The DNA pellet is then dissolved in water or any suitable buffer.

15 For the methods of the present invention, the chaotropic solution is generally an aqueous mixture of effective amounts of one or more chaotropic agents. Any chaotropic agent is useful in the practice of the present invention including but not limited to guanidine thiocyanate (GnSCN), sodium iodide, sodium perchlorate, guanidine hydrochloride, urea, hydroxides such as sodium or potassium hydroxide,
20 guanidine salt, potassium thiocyanate, formamide, and sodium chloride. Chaotropic agents include a combination of these reagents, such as a mixture of base with urea or guanidine hydrochloride.

Preferred embodiments of the present invention use aqueous solutions of 6 M guanidine hydrochloride. Best results are achieved when the chaotropic agent is

present in relatively high concentrations ranging from 1 M to 12 M. Preferred chaotropic agents for the solution include guanidine thiocyanate, guanidine hydrochloride, sodium iodide, and mixtures thereof. The concentration of chaotropes in the solution is preferably in the range of from about 1 M to about 7 M and is more preferably in the range of from about 2 M to about 5 M.

The chaotropic solution generally comprises one or more chaotropic agents and optionally contains one or more compounds selected from the group consisting of buffers or buffer solutions, solvents, surfactants, salts, preservatives, anti-oxidants, bacteriostats, solutes, reducing agents, stabilizers and mixtures thereof.

Representative acceptable buffers comprise nontoxic buffer and solution thereof, solutions that resist change of pH thereby giving stability to the components present in the solution. In another embodiment the buffer and solutions thereof can be used alone and, in a presently preferred embodiment, the buffer and solution thereof is used in combination with acceptable aqueous miscible fluids. Typical buffer aqueous solutions comprise sodium dihydrogen phosphate and disodium monohydrogen phosphate; sodium dihydrogen phosphate, disodium monohydrogen phosphate and sodium chloride; sodium carbonate, sodium monohydrogen carbonate and sodium chloride; potassium dihydrogen phosphate and sodium monohydrogen phosphate; potassium dihydrogen phosphate and sodium monohydrogen phosphate; potassium hydrogen tartrate and potassium dihydrogen phosphate; acetic acid and sodium acetate, and the like. The buffers comprise also citric acid and sodium hydroxide; potassium hydrogen phthalate and sodium hydroxide; potassium hydrogen phosphate and sodium phosphate; tris(hydroxymethyl) aminomethane and hydrochloric acid; sodium tetraborate and hydrochloric acid; glycine and hydrochloric acid;

triethanolamine and hydrochloric acid; N-tris (hydroxymethyl)methyl-2-amino sulfonic acid and sodium hydroxide, and the like. The buffer aqueous solution in another embodiment can comprise a sole component such as sodium phosphate monobasic, sodium phosphate dibasic, potassium hydrogen tartrate, potassium dihydrogen citrate, potassium hydrogen phthalate, sodium tetraborate, sodium carbonate, sodium hydrogen carbonate and mixtures thereof.

As used herein, "solvent" refers to alcohols and polar aprotic solvents. Alcohols are meant in the sense of the commonly used terminology for alcohol, preferably lower alcohols with 1 to 10 carbon atoms, more preferably methanol, ethanol, iso-propanol, n-propanol, or t-butanol, as well as glycerol, propylene glycol, ethylene glycol, polypropylene glycol, and polyethylene glycol, and most preferably ethanol or iso-propanol. Such alcohols are solvents that, when added to aqueous solution, increase the hydrophobicity of the solution by decreasing solution polarity. Polar aprotic solvents are such molecules as dimethyl sulfoxide (DMSO), dimethyl formamide (DMF), N-methylpyrrolidone (NMP), tetrahydrofuran (THF), dioxane, acetonitrile, etc., that can be used in place of or in addition to the alcohol. Preferably a lower alcohol such as ethanol, propanol or butanol, is also included, at from about 5% to about 40% by volume and more preferably at from about 15% to about 20% by volume.

Representative acceptable surfactants for the present purpose comprise anionic, cationic, amphoteric and nonionic surfactants. More specific examples of surfactants comprise sorbitan trioleate, sorbitan tristearate, propylene glycol monostearate; sorbitan sesquiolate; glycerol monostearate; sorbitan monooleate; propylene glycol monolaurate; sorbitan monostearate; diethylene glycol

monostearate; glycerol monostearate; diethylmonolaurate; sorbitan monopalmitate; sorbitan monolaurate; TRITONS; polyoxyethylene ethers; polyoxyethylene lauryl ether; polyoxyethylene sorbitan monostearate; polyoxyethylene sorbitan monooleate; polyoxyethylene sorbitan tristearate; polyoxyethylene sorbitan trioleate; 5 polyoxyethylene glycol monooleate; polyoxyethylene glycol monostearate; triethanolamine oleate; polyoxyethylene monyl phenol; polyethylene glycol monolaurate; polyoxyethylene sorbitan monolaurate; polyoxyethylene sorbitan monostearate; polyoxyethylene sorbitan monooleate; polyoxyethylene stearyl ether; polyoxyethylene oleyl ether; polyoxyethylene sorbitan monopalmitate; 10 polyoxyethylene cetyl ether; polyoxyethylene stearate; sodium oleate; potassium oleate; cetyl ethyl morpholinium ethosulfate; sodium lauryl sulfate; sodium caprylate; sodium caprate; sodium laurate; sodium myristate; sodium cholate; sodium desoxycholate; sodium dihydrocholate; tetradecyltrimethyl ammonium bromide; hexadecylpyridinium chloride; Tween 20; Tween 30; Tween 80, and mixtures thereof.

15 A compatible surfactant is generally included in an amount sufficient to promote solubilization of contaminating proteins, but at low enough concentration so that the detergent stays in solution. The preferred surfactants are non-ionic detergents such as the octoxynols and polysorbates. Examples of these are detergents with the trade names of TRITON X-100, TRITON X-114, NP 40, CONCO-NIX-100, IGEPAL 20 CA-630, AND NEUTRONYX 605. The amount of surfactant used is preferably about 0.001% to 7.5%.

Acceptable salts include, but are not limited to, those prepared from the following acids: hydrochloric, hydrobromic, sulfuric, nitric, phosphoric, maleic, acetic, salicylic, p-toluene sulfonic, tartaric, citric, methane sulfonic, formic, malonic,

succinic, naphthalene-2-sulfonic, and benzene sulfonic. Also, acceptable salts can be prepared as alkaline metal or alkaline earth salts, such as sodium, potassium, lithium or calcium salts (potassium chloride, sodium acetate, sodium nitrite, lithium chloride, and sodium bromide.). The salt concentration is generally in the range of about 0.01 M to about 1 M and preferably in the range of about 0.1 M to about 0.5 M.

As used herein, "reducing agent" refers to a compound that, in a suitable concentration in aqueous solution, maintains sulfhydryl groups so that the intra- or intermolecular disulfide bonds are chemically disrupted. Representative examples of suitable reducing agents include dithiothreitol (DTT), dithioerythritol (DTE), beta-mercaptoethanol (BME), cysteine, cysteamine, thioglycolate, glutathione, and sodium borohydride. While emphasis is placed on thiol-containing compounds, any material that is capable of the disulfide to thiol conversion without undesirable side reactions is included in this definition.

In another preferred embodiment, at least one antioxidant is added to the solutions of the present invention. The antioxidants are preferably BHA, BHT, octyl or dodecyl gallate, SO₂, for instance in the form of sodium sulfites or sodium thiosulfite, lactic acid, citric acid, tartaric acid and/or the salts thereof, vitamin C, vitamin E and uric acid and the salts thereof. Especially preferred, physiologically tolerated antioxidants are alpha -tocopherol, bilirubin, vitamin C, vitamin E, uric acid, and the salts and/or derivatives thereof. Uric acid and the salts thereof are preferably added in a concentration of from about 0.01 to about 1 wt. %, vitamin C in a concentration of from about 0.01 to 2 wt. %, and vitamin E in a concentration of from about 0.01 to about 0.1 wt. %.

The preservatives used may be formic acid in a concentration of preferably from about 0.03 to about 0.4 wt. %, acetic acid (more preferably from about 0.3 to about 3 wt. %), propionic acid, lactic acid and sorbic acid in a concentration of preferably 0.05 to 6 wt. %, especially preferred is from about 0.05 to about 12 wt. %, SO₂ (preferably from about 0.01 to about 0.6 wt. %), salicylic acid and the salts thereof (preferably from about 0.01 to about 0.5 wt. %), PBH ester (preferably from about 0.05 to about 0.6 wt. %), imidazolidinyl urea derivatives (preferably from about 0.01 to about 0.6 wt. %), chlorohexidine, NIPA-ESTER Registered TM or antibiotics.

Sterilization can be accomplished by any art-recognized technique, including but not limited to, filtration or addition of antibacterial or antifungal agents, for example, paraben, chlorobutanol, phenol, sorbic acid or thimerosal. Further, isotonic agents such as sugars or sodium chloride may be incorporated in the subject compositions.

4. An alternative use for the invention

Extra-chromosomal DNA or fragments of genomic DNA can be partially purified by conventional agarose or acrylamide gel electrophoresis on the basis of size, charge, or other properties. Since these materials are unlikely to contain RNA contaminants, no RNase treatment steps are needed. The region of the gel containing the DNA of interest is excised and solublized using an enzyme that specifically cleaves the gel matrix bonds without damaging DNA, *e.g.*, beta-agarase. After solubilization, the gel and DNA solution can be treated with the chaotropic solution of the present invention to remove gel components and other contaminants. The chaotropic solution is added to the DNA/gel solution at a ratio of about 10:1, and stored at about room

temperature for about 10 minutes. An alcohol (from about 0.3 to about 1.0 volumes relative to volume of the mixture of chaotropic solution and the DNA/gel solution) is added to the mixture in order to promote precipitation of the DNA. This mixture is centrifuged to pellet the DNA. The supernatant is removed and discarded. The DNA
5 pellet is washed with ethanol or 2-propanol to remove any residual traces of the chaotropic solution. The effective final concentration of alcohol is from about 60% to about 80%. Following centrifugation, the supernatant is removed and the DNA pellet is allowed to dry. The DNA pellet is then dissolved in water or any suitable buffer.

10 Reagent Kits

The present invention also contemplates a reagent system, typically in kit form, that can be utilized in carrying out the before-described isolation methods. The system includes, in an amount sufficient for at least one isolation, a separately packaged reagent for a process for the isolation of extra-chromosomal nucleic acids in
15 a biological sample. Instructions for use of the packaged reagent are also typically included.

As used herein, the term "package" refers to a solid matrix or material such as glass, plastic, paper, foil and the like capable of holding within fixed limits reagents of the present invention.

20 "Instructions for use" typically include a tangible expression describing the reagent concentration or at least one assay method parameter such as the relative amounts of reagent and sample to be admixed, maintenance time periods for reagent/sample admixtures, temperature, buffer conditions and the like. The packaging materials discussed herein in relation to diagnostic systems are those

customarily utilized in diagnostic systems. Such materials include glass and plastic (e.g., polyethylene, polypropylene and polycarbonate) bottles, vials, plastic and plastic-foil laminated envelopes and the like.

Specifically, the invention provides a compartmentalized kit to receive, in
5 close confinement, one or more containers which comprises: (a) a first container comprising one of the chaotropic agents of the present invention; and (b) one or more other containers comprising one or more of the following: wash reagents, lysis reagents, RNase, RNase digestion buffer, and polyacrylamide or agarose-gel solubilization solution.

10 In detail, a compartmentalized kit includes any kit in which reagents are contained in separate containers. Such containers include small glass containers, plastic containers or strips of plastic or paper. Such containers allows one to efficiently transfer reagents from one compartment to another compartment such that the samples and reagents are not cross-contaminated, and the agents or solutions of
15 each container can be added in a quantitative fashion from one compartment to another. Such containers will include a container which will accept the test sample, a container which contains the chaotropic agent used in the assay, containers which contain wash reagents (such as phosphate buffered saline, Tris-buffers, etc.), and containers which contain the reagents used to perform lysis of cells, e.g., RNase
20 digestion solution, etc.

In the following examples of the present invention, all temperatures are given on the Centigrade scale, and percentages are grams (for solids) or milliliters (for liquids) per 100 ml of water.

Example 1

1. Prepare a 100 ml culture of *E. coli* harboring the plasmid DNA of interest. Suitable culture media include sterile preparations of LB or 2XYT broth, which
5 are widely known and used in the propagation of bacteria. Typically, 100 ml of broth is placed in a sterile 500 ml culture flask. The broth is inoculated with *E. coli* harboring the plasmid of interest, and the flask is placed in a shaking incubator overnight at 37°. The following day, the culture is transferred to a centrifuge tube and the bacteria are pelleted by centrifuging for 5 minutes at about
10 1900 g. This can be done in a SORVAL™ RC5B model centrifuge, using an SLA-1000 rotor at 4000 rpm.
2. Decant the supernatant and place the tube containing the pelleted bacteria in an ice bath.
3. Resuspend the bacterial pellet in 4 ml of a solution containing 50 mM Tris-HCl
15 pH 8.0 and 10 mM EDTA, by pipetting up and down in a 10 ml serological pipet.
4. When bacterial pellet is evenly suspended, add 8 ml of a solution containing 200 mM NaOH and 1% SDS. Gently mix by inverting the capped tube 15-30 times. Care must be taken not to shake the mixture. Store on ice 5 minutes.
5. Add 6 ml of a solution containing 3 M potassium acetate pH 5.5. Gently mix by
20 inverting the tube 15-30 times. Care must be taken to not shake the mixture. Store on ice for 10 minutes.
6. Separate plasmid DNA from contaminating chromosomal DNA and bacterial debris by centrifuging at 22000 g for 20 minutes at 4°. This can be done in a SORVAL™ RC5B model centrifuge, using an SS34 rotor at 13,500 rpm.

7. Transfer the supernatant to a clean 250 ml centrifuge bottle and add 40 ml ice-cold 95% ethanol. Vigorously swirl mixture by hand, and store on ice for 10 minutes.
8. Centrifuge at about 11000 g for 15 minutes at 4°. This can be done in a SORVAL™ RC5B model centrifuge, using an SLA-1000 rotor at 9,000 rpm.
- 5 9. Decant liquid. Close observation of the tube will reveal the location of a large white pellet. This is the crude isolate of nucleic acids, containing plasmid DNA, bacterial RNA, and residual protein contaminants. Carefully pipet away any excess liquid in the tube, but do not disturb the pellet.
- 10 11. Resuspend pellets in 1 ml of a solution containing 10 mM Tris-HCl pH 7.5 and 50 ug/ml RNase A.. Transfer the solution to a 1.5 ml microcentrifuge tube such as an Eppendorf™ tube. Incubate in a water bath at 37° for 10 minutes.
12. Spin down excess debris. Place the microcentrifuge tube in a microcentrifuge at maximum speed (typically 12,000-15,000 rpm) for 5-10 minutes at 4°.
- 15 13. Transfer the supernatant to a clean 30 ml centrifuge tube. Add 10 ml of a chaotropic solution containing 4M guanadine isothiocyanate, 0.05 M TRIZMA Base pH 9.2, 0.3 M sodium acetate, 0.1% TRITON X-114, and 17% 2-propanol. Mix by VORTEX® agitation for 5-10 seconds, and store at room temperature for 10 minutes.
- 20 14. Add 7 ml of ice-cold 95% ethanol, mix by VORTEX® agitation 5-10 seconds, and store on ice for 10 minutes.
15. Centrifuge at about 27000 g for 20 minutes at 4°. This can be done in a SORVAL™ RC5B model centrifuge, using an SS34 rotor at 15,000 rpm.

16. Decant the supernatant. Wash the pellet once by adding about 5 ml of 70% ethanol so that the liquid covers the pellet, mixing by VORTEX™ agitation for 10-20 seconds, and then recentrifuging in an SS34 rotor at 15,000 rpm for 10-15 minutes.
- 5 17. Use a micropipeter to carefully remove as much supernatant as possible. Leave the cap of the centrifuge tube open, and allow the DNA pellet to air dry. The pellet is usually dry after about 10-30 minutes.
18. Dissolve the DNA pellet by adding 500 µl 10 mM Tris, pH 7.8 to the centrifuge tube, and transfer the dissolved DNA to a 1.5 ml Eppendorf™ tube.
- 10 19. Place the tube in a 55° heat block for 10 minutes to promote solubilization. Agitate the solution briefly using a VORTEX™ mixer.
20. Spin for 10 minutes at maximum speed in a microcentrifuge at 4° to remove any residual contaminants from the DNA solution.
21. Transfer DNA solution to a clean 1.5 ml microcentrifuge tube for long-term
15 storage.

Example 2

1. Prepare a lysate from bacteria harboring the extra-chromosomal DNA of interest. This can be done following the protocol outlined on pp 1.21-1.52 of *Molecular Cloning, A Laboratory Manual* Second Edition. Sambrook, J.; Fritsch, E.F.;
20 Maniatis, T. Cold Spring Harbor Laboratory Press, 1989.
2. Transfer the lysate to a centrifuge tube which can hold a volume equal to or greater than 3.5 times the volume of lysate. Add 2.5 parts of 95% ethanol to 1 part lysate. Swirl vigorously by hand to mix thoroughly.

3. Centrifuge at about 21000 g for 20 minutes to the nucleic acid pellet. This can be done in a SORVAL™ RC5B model centrifuge, using an SS34 rotor at 13,500 rpm. Decant supernatant and allow the pellet to air dry.
4. Resuspend the pellet in a solution of 10 mM Tris-HCl, pH 7.5 in the presence of
5 RNase A (50 ug/ml) and transfer to an 1.5 ml microfuge tube. Incubate the lysate at 37° heat block for 10 minutes. Sediment contaminating material by centrifuging the microfuge tube at maximum speed in a microcentrifuge for 10 minutes at 4°.
5. Transfer the supernatant to a clean 30 ml centrifuge tube. Treat the supernatant
10 with 10 ml of a chaotropic solution containing 5 M guanadine isothiocyanate, 0.05 M TRIZMA Base pH 9.2, 0.1 M sodium acetate, 0.1% NP-40, and 15 % 2-propanol. Mix by VORTEX® agitation for 5-10 seconds, and store at room temperature for 10 minutes.
6. Add 7 ml of ice-cold 95% ethanol, mix by VORTEX™ agitation 5-10 seconds,
15 and store on ice for 10 minutes.
7. Centrifuge at about 27000 g for 20 minutes at 4°. This can be done in a SORVAL™ RC5B model centrifuge, using an SS34 rotor at 15,000 rpm.
8. Decant the supernatant. Wash the pellet once by adding about 5 ml of 70% ethanol so that the liquid covers the pellet, mixing by VORTEX™ agitation for
20 10-20 seconds, and then recentrifuging in a microcentrifuge at maximum speed for 10-15 minutes.
9. Use a micropipeter to carefully remove as much supernatant as possible. Leave the cap of the microcentrifuge tube open, and allow the DNA pellet to air dry. The pellet is usually dry after about 10-30 minutes.

10. Dissolve the DNA pellet by adding 500 μ l 10 mM Tris, pH 7.8 to the microcentrifuge tube.
11. Place the tube in a 55° heat block for 10 minutes to promote solubilization. Agitate the solution briefly using a VORTEX™ mixer.
- 5 12. Spin for 10 minutes at maximum speed in a microcentrifuge at 4° to remove any residual contaminants from the DNA solution.
13. Transfer DNA solution to a clean 1.5 ml microcentrifuge tube for long-term storage.

10 Example 3

Prepare a lysate from 500 ml of bacterial culture harboring the BAC DNA of interest. This can be done following the protocol outlined on pp 1.21-1.52 of *Molecular Cloning, A Laboratory Manual* Second Edition. Sambrook, J.; Fritsch, E.F.; Maniatis, T. Cold Spring Harbor Laboratory Press, 1989. Because only one copy of

15 a BAC will be present in each bacterium, as opposed to plasmid DNA with anywhere from 20-300 copies per bacterium, much larger culture volumes is required to isolate a useful amount of DNA. While the amounts of reagents needed for the alkaline lysis and neutralization steps will be proportionately increased, the precipitated pellet obtained from the lysate is resuspended in about 1-2 ml of 10 mM Tris-HCl pH 7.5

20 buffer for the RNase treatment step.

1. Transfer the lysate to a centrifuge tube that can hold a volume equal to or greater than 3.5 times the volume of lysate. Add 2.5 parts of 95% ethanol to 1 part lysate. Swirl vigorously by hand to mix thoroughly.

2. Centrifuge at about 11000 g for 15 minutes to pellet crude nucleic acids. This can be done in a SORVAL™ RC5B model centrifuge, using an SLA-1000 rotor at 9,000 rpm. Decant supernatant.
3. Resuspend the pellet in a solution of 10 mM Tris-HCl, pH 7.5 containing RNase
5 A and transfer to one or several 1.5 ml microfuge tubes. Because a greater amount of culture volume is needed to isolate BAC DNA, there will be a corresponding increase in the amount of RNA contamination. Add RNase A to the lysate such that the final concentration of RNase A is 200 ug/ml. Incubate the lysate at 37° on a heat block for 1 hour. Pellet contaminating material away from
10 the extra-chromosomal DNA by centrifuging at maximum speed in a microcentrifuge at 4°.
4. Transfer the RNase-treated supernatant to a 30 ml centrifuge tube. Treat the supernatant with 10 ml of a chaotropic solution containing 4 M guanidine isothiocyanate, 0.05 M TRIZMA Base pH 10, 0.5 M sodium acetate, 0.45%
15 TRITON X-100, and 20% ethanol. Mix by VORTEX™ agitation for 5-10 seconds, and store at room temperature for 10 minutes.
5. Add 7 ml of ice-cold 95% ethanol, mix by VORTEX™ agitation 5-10 seconds, and store on ice for 10 minutes.
6. Centrifuge at about 27000 g for 20 minutes to pellet crude DNA. This can be
20 done in a SORVAL™ RC5B model centrifuge, using an SS34 rotor at 15,000 rpm.
7. Decant the supernatant. Wash the pellet once by adding about 5 ml of 70% ethanol so that the liquid covers the pellet, mixing by VORTEX™ agitation for 10-20 seconds, and then recentrifuging for 10-15 minutes.

8. Use a micropipeter to carefully remove as much supernatant as possible. Leave the cap of the centrifuge tube open, and allow the DNA pellet to air dry. The pellet is usually dry after about 10-30 minutes.
9. Dissolve the DNA pellet by adding 500 μ l 10 mM Tris, pH 7.8 and transfer the dissolved DNA to a microcentrifuge tube.
10. Place the tube in a 55° heat block for 10 minutes to promote solubilization. Agitate the solution briefly using a VORTEX ® mixer.
11. Spin for 10 minutes at maximum speed in a microcentrifuge at 4° to remove any residual contaminants from the DNA solution.
12. Transfer DNA solution to a clean 1.5 ml microcentrifuge tube for long-term storage.

Example 4

1. Prepare a 30 ml culture of *E. coli* harboring the plasmid DNA of interest. Suitable culture media include sterile preparations of LB or 2XYT broth, which are widely known and used in the propagation of bacteria. Typically, 30 ml of broth is placed in a sterile 150 ml culture flask. The broth is inoculated with *E. coli* harboring the plasmid of interest, and the flask is placed in a shaking incubator overnight at 37°. The following day, the culture is transferred to a 50 ml FALCON® tube and the bacteria are pelleted by centrifuging for 5 minutes at 1500 g. This can be done in a Beckman centrifuge, using an JA20 rotor at 2000 rpm.
2. Decant the supernatant and place the tube containing the pelleted bacteria in an ice bath.

3. Resuspend the bacterial pellet in 2 ml of a solution containing 50 mM Tris-HCl pH 8.0 and 10 mM EDTA, by pipeting up and down in a 10 serological pipet.
4. When bacterial pellet is evenly suspended, add 4 ml of a solution containing 200 mM NaOH and 1% SDS. Gently mix by inverting the capped tube 15-30 times.
- 5 Care must be taken to not shake the mixture. Store on ice 5 minutes.
5. Add 3 ml of a solution containing 3 M potassium acetate pH 5.5. Gently mix by inverting the tube 15-30 times. Care must be taken to not shake the mixture. Store on ice for 10 minutes.
6. Separate plasmid DNA from contaminating chromosomal DNA and bacterial
10 debris by centrifuging at about 8000g for 30 minutes at 4°. This can be done in a Beckman centrifuge, using an JA20 rotor at 10,000 rpm.
7. Transfer the supernatant to a clean 50 ml FALCON tube and add 25 ml ice-cold 95% ethanol. Vigorously swirl mixture by hand, and store on ice for 10 minutes.
8. Centrifuge at 8000 g for 30 minutes at 4°. This can be done in a Beckman
15 centrifuge, using an JA20 rotor at 10,000 rpm.
9. Decant liquid. Close observation of the tube will reveal the location of a small white pellet. This is the crude isolate of nucleic acids, containing plasmid DNA, bacterial RNA, and residual protein contaminants. Carefully pipet away any excess liquid in the tube, but do not disturb the pellet.
- 20 10. Leave the cap off the tube and allow the pellet to air dry for 5-10 minutes.
11. Resuspend pellet in 1 ml of a solution containing 10 mM Tris-HCl pH 7.5 and 50 ug/ml RNase A.. Transfer the solution to a 1.5 ml microcentrifuge tube such as an Eppendorf™ tube. Incubate in a water bath at 37° for 10 minutes.

12. Spin down excess debris. Place the microcentrifuge tube in a microcentrifuge at maximum speed (typically 12,000-15,000 rpm) for 5-10 minutes at 4°.
13. Transfer the supernatant to a clean 50 ml FALCON™ tube. Add 5 ml of a chaotropic solution containing 3 M guanidine isothiocyanate, 0.1 M guanidine hydrochloride, 0.05 M TRIZMA Base pH 8.5, 0.2 M sodium acetate, 0.5% TRITON X-100, and 17% 2-propanol. Mix by VORTEX™ agitation for 5-10 seconds, and store at room temperature for 10 minutes.
14. Add 3.5 ml of ice-cold 95% ethanol, mix by VORTEX™ agitation 5-10 seconds, and store on ice for 10 minutes.
15. Centrifuge at 8000 g for 30 minutes at 4°. This can be done in a Beckman centrifuge, using an JA20 rotor at 10,000 rpm.
16. Decant the supernatant. Wash the pellet once by adding about 3 ml of 70% ethanol so that the liquid covers the pellet, mixing by VORTEX™ agitation for 10-20 seconds, and then recentrifuge for 10-15 minutes.
17. Use a micropipeter to carefully remove as much supernatant as possible. Leave the cap of the 50 ml FALCON™ tube open, and allow the DNA pellet to air dry. The pellet is usually dry after about 20-30 minutes.
18. Dissolve the DNA pellet by adding 500 µl 10 mM Tris, pH 7.8 and transfer the dissolved DNA to a microcentrifuge tube.
19. Place the tube in a 55° heat block for 10 minutes to promote solubilization. Agitate the solution briefly using a VORTEX™ mixer.
20. Spin for 10 minutes at maximum speed in a microcentrifuge at 4° to remove any residual contaminants from the DNA solution.

21. Transfer DNA solution to a clean 1.5 ml microcentrifuge tube for long-term storage.

Example 5.

- 5 1. Prepare a 2 ml culture of *E. coli* harboring the plasmid DNA of interest. Suitable culture media include sterile preparations of LB or 2XYT broth, which are widely known and used in the propagation of bacteria. Typically, 2 ml of broth is placed in a sterile 5 ml culture tube. The broth is inoculated with *E. coli* harboring the plasmid of interest, and the flask is placed in a shaking incubator overnight at 37°.
- 10 The following day, the bacteria are pelleted by centrifuging for 2 minutes at maximum speed (12000-15000 rpm) in a microcentrifuge at room temperature.
2. Decant the supernatant and place the tube containing the pelleted bacteria in an ice bath.
3. Resuspend the bacterial pellet in 100 µl of a solution containing 50 mM Tris-HCl
- 15 pH 8.0 and 10 mM EDTA.
4. When bacterial pellet is evenly suspended, add 200 µl of a solution containing 200 mM NaOH and 1% SDS. Gently mix by inverting the capped tube 15-30 times. Care must be taken to not shake the mixture. Store on ice 5 minutes.
5. Add 200 µl of a solution containing 3 M potassium acetate pH 5.5. Gently mix by
- 20 inverting the tube 15-30 times. Care must be taken to not shake the mixture. Store on ice for 10 minutes.
6. Separate plasmid DNA from contaminating chromosomal DNA and bacterial debris by centrifuging at maximum speed (12000-15000 rpm) in a microcentrifuge at 4° for about 10 minutes.

7. Transfer the supernatant to a clean 1.5 ml microcentrifuge tube and add 900 μ l ice-cold 95% ethanol. Mix vigorously, and store on ice for 10 minutes.
8. Centrifuge at maximum speed (12000-15000 rpm) in a microcentrifuge at 4° for about 20 minutes
- 5 9. Decant liquid. Close observation of the tube will reveal the location of a small white pellet. This is the crude isolate of nucleic acids, containing plasmid DNA, bacterial RNA, and residual protein contaminants. Carefully pipet away any excess liquid in the tube, but do not disturb the pellet.
- 10 10. Leave the cap off the tube and allow the pellet to air dry for 5-10 minutes.
- 10 11. Resuspend pellet in 50 μ l of a solution containing 10 mM Tris-HCl pH 7.5 and 50 μ g/ml RNase A. Incubate in a water bath at 37° for 10 minutes.
12. Spin down excess debris. Place the microcentrifuge tube in a microcentrifuge at maximum speed (typically 12,000-15,000 rpm) for 5-10 minutes at 4°.
13. Transfer the supernatant to a clean 1.5 ml microcentrifuge tube and add 700 μ l of
15 a chaotropic solution containing 3M guanidine isothiocyanate, 0.05 M TRIZMA Base pH 9.0, 0.4 M sodium acetate, 0.3% TRITON X-100, and 17% ethanol. Mix by VORTEX agitation for 5-10 seconds, and store at room temperature for 10 minutes.
14. Add 350 μ l of ice-cold 95% ethanol, mix by VORTEX agitation 5-10 seconds,
20 and store on ice for 10 minutes
15. Place the microcentrifuge tube in a microcentrifuge at maximum speed (typically 12,000-15,000 rpm) for 5-10 minutes at 4°.
16. Decant the supernatant. Wash the pellet once by adding about 300 μ l of 70% ethanol so that the liquid covers the pellet, mixing by VORTEX agitation for 10-

20 seconds, and then recentrifuging in a microcentrifuge at maximum speed for 10-15 minutes.

17. Use a micropipeter to carefully remove as much supernatant as possible. Leave the cap of the microcentrifuge tube open, and allow the DNA pellet to air dry.

5 The pellet is usually dry after about 10-30 minutes.

18. Dissolve the DNA pellet by adding 90 μ l 10 mM Tris, pH 7.8 to the microcentrifuge tube.

19. Place the tube in a 55° heat block for 10 minutes to promote solubilization. Agitate the solution briefly using a VORTEX mixer.

10 20. Spin for 10 minutes at maximum speed in a microcentrifuge at 4° to remove any residual contaminants from the DNA solution.

21. Transfer DNA solution to a clean 1.5 ml microcentrifuge tube for long-term storage.

15 Example 6

1. Fragments of chromosomal DNA are partially purified by conventional agarose gel electrophoresis on the basis of size.

2. The region of the gel containing the DNA of interest is excised and placed in a microcentrifuge tube containing 3.0:0.2 Tris EDTA buffer.

20 3. The tube is placed in a 55° heat block until the gel is melted for about 10 minutes.

4. Invert tube several times to mix thoroughly. Add 700 μ l of a chaotropic solution containing 4M guanadine isothiocyanate, 0.05 M TRIZMA Base pH 9.5, 0.3 M sodium acetate, 0.4% TRITON X-100, and 16% 2-propanol. Store at about room

temperature for 10 minutes. Since these materials are unlikely to contain RNA contaminants, no precipitation or RNase treatment steps are needed.

5. Precipitate the DNA by adding 500 μ l ice-cold 95% ethanol and storing on ice for 10 minutes. Centrifuge the mixture of the chaotropic solution and the DNA/gel solution is to pellet the DNA. This can be done by placing the microcentrifuge tube in a microcentrifuge at maximum speed (typically 12,000-15,000 rpm) for 5-10 minutes at 4°.
6. Remove and discard supernatant. Wash the DNA pellet with 500 μ l of 70% ethanol to remove any residual traces of the chaotropic solution and then re-centrifuge at maximum speed.
7. Following centrifugation, remove the supernatant and allow the DNA pellet to dry.
8. The DNA pellet is then dissolved in water or any suitable buffer.

Claims.

What is claimed is:

1. A method of isolating extrachromosomal nucleic acids in a substantially purified
5 form, the method comprising the steps of:
 - a) homogenizing a biological sample;
 - b) removing substantially all of the chromosomal nucleic acids from the biological
sample;
 - c) contacting the biological sample containing extra-chromosomal nucleic acids
10 with an effective amount of a chaotropic solution under conditions which permit the
nucleic acids in the supernatant to precipitate; and
 - d) recovering the extrachromosomal nucleic acids.
2. The method of claim 1, further comprising the step, prior to step c), of contacting
15 the biological sample with a ribonuclease enzyme under conditions sufficient to
subject the sample to endonucleolytic cleavage of RNA.
3. The method of claim 2, further comprising the step, prior to step c), of precipitating
said extrachromosomal nucleic acids and resuspending the nucleic acids in buffer.
20
4. The method of claim 3, wherein the homogenizing of the biological sample is by
incomplete lysis.
5. The method of claim 4, wherein the incomplete lysis is by alkaline lysis.

6. The method of claim 5, wherein the chromosomal nucleic acids are removed by the method selected from the group consisting of precipitation, sedimentation, filtration, and centrifugation.
- 5
7. The method of claim 6, further comprising the step of washing said recovered extrachromosomal nucleic acids
8. The method of claim 7, wherein the washing is by addition of an alcohol and water
- 10 solution.
9. The method of claim 6, wherein the biological sample is a cell system selected from the group consisting of bacteria, yeast, insect, plant and mammalian.
- 15 10. The method of claim 9, wherein the cell system contains one or more extracellular nucleic acid sources selected from the group consisting of recombinant bacteriophage, plasmid, cosmid, yeast expression vector, viral expression vector, and retrovirus.
- 20 11. The method of claim 5, wherein the recovery of said nucleic acids from said solution is by adding an effective amount of organic solvent thereto, and recovering the precipitated nucleic acids.
12. The method of claim 11, wherein said organic solvent is selected from the group

consisting of lower alcohols, acetone, polyethylene glycol and dimethylsulfoxide.

13. The method of claim 6, wherein said chaotropic solution comprises:

- a) effective amounts of at least one chaotropic agent, and
- 5 b) a buffer present in an amount sufficient to maintain said solution at an alkaline pH.

14. The method of claim 6, wherein said chaotropic agent is selected from the group consisting of guanidine thiocyanate, sodium iodide, sodium perchlorate, guanidine
10 hydrochloride, urea, sodium hydroxide, potassium hydroxide, guanidine salt, potassium thiocyanate, formamide, and sodium chloride, sodium iodide and mixtures thereof.

15. The method of claim 14, wherein said chaotropic agent is present at a
15 concentration in the range of about 1M-7M.

16. The method of claim 14, wherein said chaotropic agent is present at a concentration in the range of about 2M-5M.

20 17. The method of claim 16, wherein said buffer is selected from the group consisting of sodium dihydrogen phosphate and disodium monohydrogen phosphate; sodium dihydrogen phosphate, disodium monohydrogen phosphate and sodium chloride; sodium carbonate, sodium monohydrogen carbonate and sodium chloride; potassium dihydrogen phosphate and sodium monohydrogen phosphate; potassium dihydrogen

phosphate and sodium monohydrogen phosphate; potassium hydrogen tartrate and potassium dihydrogen phosphate; acetic acid and sodium acetate, citric acid and sodium hydroxide; potassium hydrogen phthalate and sodium hydroxide; potassium hydrogen phosphate and sodium phosphate; tris(hydroxymethyl) aminomethane and
5 hydrochloric acid; sodium tetraborate and hydrochloric acid; glycine and hydrochloric acid; triethanolamine and hydrochloric acid; N-tris (hydroxymethyl)methyl-2-amino sulfonic acid and sodium hydroxide; sodium phosphate monobasic, sodium phosphate dibasic, potassium hydrogen tartrate, potassium dihydrogen citrate, potassium hydrogen phthalate, sodium tetraborate, sodium carbonate, sodium hydrogen
10 carbonate and mixtures thereof.

18. The method of claim 17, wherein said buffer is present in an amount sufficient to maintain the pH of said solution in the range of about 7 to about 12.

15 19. The method of claim 17, wherein said buffer is present in an amount sufficient to maintain the pH of said solution in the range of about 7 to about 10.

20. The method of claim 17, wherein said buffer is present in an amount sufficient to maintain the pH of said solution in the range of about 7 to about 8.

20

21. The method of claim 17, wherein said chaotropic solution further comprising one or more compounds selected from the group consisting of solvents, surfactants, detergents, salts, chelating agents, preservatives, anti-oxidants, bacteriostats, solutes, reducing agents, stabilizers and mixtures thereof

22. The method of claim 21, wherein said reducing agent is selected from the group consisting of dithiothreitol (DTT), dithioerythritol (DTE), beta-mercaptoethanol (BME), cysteine, cysteamine, thioglycolate, glutathione, sodium borohydride and
5 mixtures thereof.
23. The method of claim 21, wherein said reducing agent is beta-mercaptoethanol.
24. The method of claim 21, wherein said salt is selected from the group consisting of
10 sodium chloride, potassium chloride, ammonium chloride, sodium acetate, sodium nitrate, lithium chloride, sodium bromide and mixtures thereof.
25. The method of claim 24, wherein said salt is present in the range from about 0.01 M to about 1 M. and preferably in the range of about 0.1 M to about 0.5 M.
15
26. The method of claim 24, wherein said salt is present in the range from about 0.1 M to about 0.5 M.
27. The method of claim 21, wherein said chelating agent is selected from the group
20 consisting of ethylenediamine tetraacetic acid, citric acid and mixtures thereof.
28. The method of claim 27, wherein said chelating agent is present in the range of about 0.05% to about 0.50%.

29. The method of claim 21, wherein said detergent is selected from the group consisting of sorbitan trioleate, sorbitan tristearate, propylene glycol monostearate; sorbitan sesquiolate; glycerol monostearate; sorbitan monooleate; propylene glycol monolaurate; sorbitan monostearate; diethylene glycol monostearate; glycerol monostearate; diethylmonolaurate; sorbitan monopalmitate; sorbitan monolaurate; TRITONs; polyoxyethylene ethers; polyoxyethylene lauryl ether; polyoxyethylene sorbitan monostearate; polyoxyethylene sorbitan monooleate; polyoxyethylene sorbitan tristearate; polyoxyethylene sorbitan trioleate; polyoxyethylene glycol monooleate; polyoxyethylene glycol monostearate; triethanolamine oleate; polyoxyethylene monyl phenol; polyethylene glycol monolaurate; polyoxyethylene sorbitan monolaurate; polyoxyethylene sorbitan monostearate; polyoxyethylene sorbitan monooleate; polyoxyethylene stearyl ether; polyoxyethylene oleyl ether; polyoxyethylene sorbitan monopalmitate; polyoxyethylene cetyl ether; polyoxyethylene stearate; sodium oleate; potassium oleate; cetyl ethyl morpholinium ethosulfate; sodium lauryl sulfate; sodium caprylate; sodium caprate; sodium laurate; sodium myristate; sodium cholate; sodium desoxycholate; sodium dihydrocholate; tetradecyltrimethyl ammonium bromide; hexadecylpyridinium chloride; Tween 20; Tween 30; Tween 80, and mixtures thereof.
30. The method of claim 29, wherein said detergent is present in the range of about 0.001% to about 7.5%.
31. The method of claim 29, wherein said detergent is TRITON X-114®.

32. The method of claim 21, wherein said solvent is selected from the group consisting of lower alcohols, acetone, polyethylene glycol, dimethylsulfoxide, and mixtures thereof..
- 5 33. The method of claim 32, wherein said lower alcohol is selected from the group consisting of methanol, ethanol, iso-propanol, n-propanol, t-butanol, glycerol, propylene glycol, ethylene glycol, polypropylene glycol, polyethylene glycol, and mixtures thereof.
- 10 34. The method of claim 32, wherein said solvent is present in the range of from about 5% to about 40% by volume.
35. The method of claim 32, wherein said solvent is present in the range of from about 15% to about 20% by volume.
- 15 36. The method of claim 35, wherein said solvent is iso-propanol.
37. The method of claim 21, wherein said antioxidant is selected from the group consisting of BHA, BHT, octyl gallate, dodecyl gallate, sodium sulfite, sodium
- 20 thiosulfite, lactic acid, citric acid, tartaric acid, vitamin C, vitamin E, uric acid and the salts and mixtures thereof.
38. A method of isolating nucleic acids in a substantially purified form, said method comprising the steps of:

- a) homogenizing a biological sample containing nucleic acids of interest;
- b) partially purifying the nucleic acids from the biological sample by adhering the nucleic acids of interest to a retaining means;
- c) contacting the gel containing the sample with an effective amount of a chaotropic solution under conditions which permit the nucleic acids to precipitate; and
- d) recovering the nucleic acids.

39. The method of claim 38, wherein the retaining means is a solid support matrix

40. The method of claim 39, wherein the solid support matrix is an electrophoresis gel

41. The method of claim 39, wherein the gel is agarose

42. The method of claim 39, wherein the gel is acrylamide

43. The method of claim 40, wherein the region of the gel containing the DNA of interest is excised and solublized prior to contacting the sample with a chaotropic solution.

44. The method of claim 41, wherein the gel is solublized using an enzyme that specifically cleaves the gel matrix bonds without damaging nucleic acids.

45. The method of claim 44, wherein the enzyme is beta-agarase.

46. The method of claim 40, wherein the recovery of said nucleic acids from said solution is by adding an effective amount of organic solvent thereto, and recovering the precipitated nucleic acids.

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47. The method of claim 46, wherein said organic solvent is selected from the group consisting of lower alcohols, acetone, polyethylene glycol and dimethylsulfoxide.

48. The method of claim 47, wherein said chaotropic solution comprises:

- 10 a) effective amounts of at least one chaotropic agent, and
 b) a buffer present in an amount sufficient to maintain said solution at an alkaline pH.

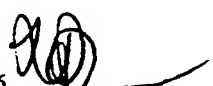
49. The method of claim 48, wherein said chaotropic agent is selected from the group
15 consisting of guanidine thiocyanate, sodium iodide, sodium perchlorate, guanidine hydrochloride, urea, sodium hydroxide, potassium hydroxide, guanidine salt, potassium thiocyanate, formamide, and sodium chloride, sodium iodide and mixtures thereof.

20 50. The method of claim 49, wherein said chaotropic agent is present at a concentration in the range of about 1M-7M.

51. The method of claim 49, wherein said chaotropic agent is present at a concentration in the range of about 2M-5M.

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US00/16712

A. CLASSIFICATION OF SUBJECT MATTER IPC(7) :C12P 19/34; C12N 9/00, 13/00 US CL :435/91.1, 173.7, 183 According to International Patent Classification (IPC) or to both national classification and IPC														
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) U.S. : 435/91.1, 173.7, 183 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) Please See Extra Sheet.														
C. DOCUMENTS CONSIDERED TO BE RELEVANT														
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.												
A	MILLER, H. 'Practical Aspects of Preparing Phage and Plasmid DNA: Growth, Maintenance, and Storage of Bacteria and Bacteriophage.' In: Guide to Molecular Cloning Techniques. Edited by Berger et al. San Diego: Academic Press, 1987, Vol. 152, pages 145-170.	1-51												
A	1996/1997 New England BioLabs Catalog. New England Biolabs, Inc. 1996, page 91.	38-51												
A	US 5,834,303 A (M. FUJISHIRO) 10 November 1998 (10.11.98), see entire document.	1-51												
A, P	US 6,027,750 A (GAUTSCH et al) 22 February 2000 (22.02.00), see entire document.	1-51												
<input type="checkbox"/> Further documents are listed in the continuation of Box C. <input type="checkbox"/> See patent family annex.														
<table border="0"> <tr> <td>* Special categories of cited documents:</td> <td>*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</td> </tr> <tr> <td>*A* document defining the general state of the art which is not considered to be of particular relevance</td> <td>*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone</td> </tr> <tr> <td>*E* earlier document published on or after the international filing date</td> <td>*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</td> </tr> <tr> <td>*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</td> <td>*G* document member of the same patent family</td> </tr> <tr> <td>*O* document referring to an oral disclosure, use, exhibition or other means</td> <td></td> </tr> <tr> <td>*P* document published prior to the international filing date but later than the priority date claimed</td> <td></td> </tr> </table>			* Special categories of cited documents:	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention	*A* document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone	*E* earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art	*L* document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*G* document member of the same patent family	*O* document referring to an oral disclosure, use, exhibition or other means		*P* document published prior to the international filing date but later than the priority date claimed	
* Special categories of cited documents:	*T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention													
A document defining the general state of the art which is not considered to be of particular relevance	*X* document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone													
E earlier document published on or after the international filing date	*Y* document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art													
L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)	*G* document member of the same patent family													
O document referring to an oral disclosure, use, exhibition or other means														
P document published prior to the international filing date but later than the priority date claimed														
Date of the actual completion of the international search 24 AUGUST 2000		Date of mailing of the international search report 13 OCT 2000												
Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT Washington, D.C. 20231 Facsimile No. (703) 305-3230		Authorized officer GERALD G. LEFFERS JR.  Telephone No. (703) 308-0196												

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US00/16712

B. FIELDS SEARCHED

Electronic data bases consulted (Name of data base and where practicable terms used):

STN: BIOSIS EMBASE MEDLINE CAPLUS; EAST: DERWENT EPO JPO USPAT
search terms: inventor names, extrachromosomal, DNA, plasmid, nucleic acid, purification, isolation, beta-agarase, gelase, guanidium, chaotropic, thiocyanate